

## NANOCOMPOSITES IN THE $ZrO_2$ - $Al_2O_3$ -WC SYSTEM

Z. Pędzich<sup>1</sup>, W. Maziarz<sup>2</sup>

<sup>1</sup> AGH – University of Science and Technology, Faculty of Materials Science and Ceramics, Department of Advanced Ceramics, 30 Mickiewicza Av., 30-059 Krakow, Poland

<sup>2</sup> Institute of Metallurgy and Materials Science of Polish Academy of Sciences, 25 Reymonta Str., 30-058 Krakow, Poland

Received 01.12.2010

Accepted 03.02.2011

*Corresponding author: Zbigniew Pędzich, AGH – University of Science and Technology, Faculty of Materials Science and Ceramics, Department of Advanced Ceramics, 30 Mickiewicza Av., 30-059 Krakow, Poland, pedzich@agh.edu.pl, tel. +48 2 617 2397, fax +48 12 633 4630*

### Abstract

Particulate composite in the  $Al_2O_3$ - $ZrO_2$ -WC carbide system shows promising properties in structural applications especially as wear resistant parts. Mentioned materials show the unique microstructure of sintered bodies composed of three continuous polycrystalline subsystems mutually penetrating each other. Despite of mechanical properties of constituent phases, its properties depends on the residual stress state and the microstructure. Especially important is the interphase grain boundaries state. Significant differences of thermal properties of constituent phases create during composite manufacturing meaningful residual stresses (locally exceeded 1.5 MPa). It can influence the interphase boundaries structure. The Paper presents results of investigation of high-resolution transmission electron microscopy (HREM) on the different type grain boundaries structure and its chemical composition.

**Keywords:** composites, microstructure, zirconia, alumina, tungsten carbide

### 1 Introduction

Sintered composites with alumina or/and zirconia matrix were identified as relatively low cost materials showing very good abrasive wear resistance [1, 2]. Fabrication of the dense composite materials using typical powder compacting technology and following sintering process meets many barriers connected with each step of this technology. Application of submicro- or nanosized powders increases difficulties occurring during manufacturing [3, 4].

The first task for applied technology is to assure the uniformity of different phase distribution in prepared composite powder. The process must eliminate the possibility of one phase grains aggregation. If it is performed with success, sintering could provide to homogeneous composite body with properties significantly improved in relation to pure matrix material [5–7].

Usually it is profitable for sintering process to avoid of chemical reactions, nevertheless sintering must be provided at temperatures, which assure the effective mass transport in sintered phases.

Additionally, after sintering process, during cooling, in dense multiphase bodies residual stresses could be created due to thermal expansion coefficient mismatch [8, 9].

This work shows the results of detail investigation of the fine-grained microstructure of three-phase composite containing comparable amount of each phase. The special care was put to

describe grain boundaries between the grains of different phases (i. e. alumina-carbide or zirconia-carbide) because the state of interphase boundaries strongly influenced the composite properties [10-12].

## 2 Experimental

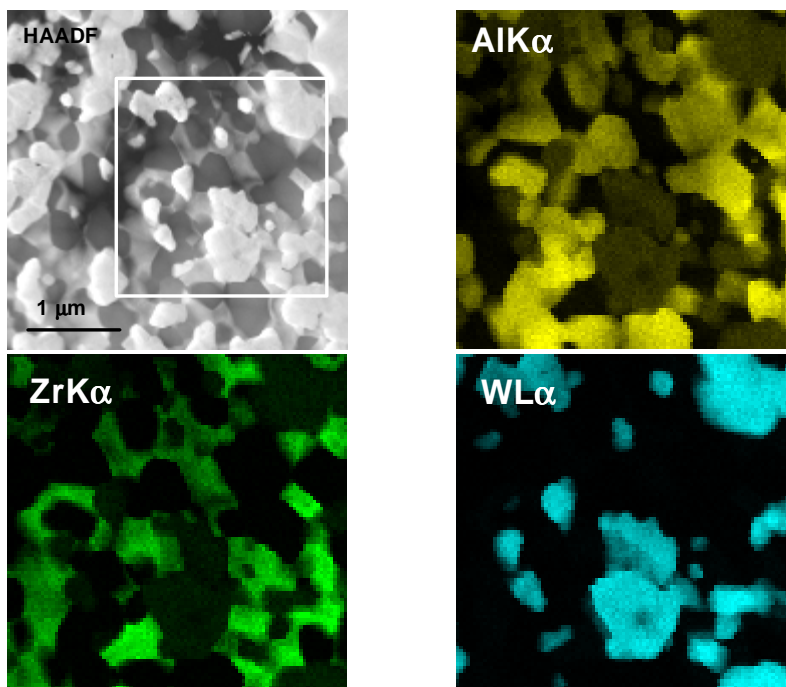
$\text{Al}_2\text{O}_3\text{-ZrO}_2\text{-WC}$  composite powder was prepared from commercial powders ( $\alpha$ -alumina: TM-DAR, Taimei Chemicals; yttria stabilized zirconia: 3Y-TZ, Tosoh and tungsten carbide: Baildonit) by wet (ethanol) mixing in attritor. The volume proportion of each phase was 1:1:1. Sintering of composite bodies was conducted in the Thermal Technology hot-press device at  $1650^\circ\text{C}$  (with 45 min. soaking time) in argon atmosphere under 25 MPa pressure. Density of sintered bodies was  $8.46\text{ g/cm}^3$  (98.5% of theoretical value).

The structure investigations were performed in Tecnai G2 FEG microscope dedicated for high resolution (HREM) observations. The thin foils were prepared by cutting thin slices of compacts using diamond saw and followed dimpling and ion milling using Gatan equipments. Finally the samples were covered by carbon thin layer in order to assure their conductivity.

The chemical analyses both point and line scans were performed in the HAADF/STEM mode using a energy dispersive X-ray spectrometry (EDS).

## 3 Results

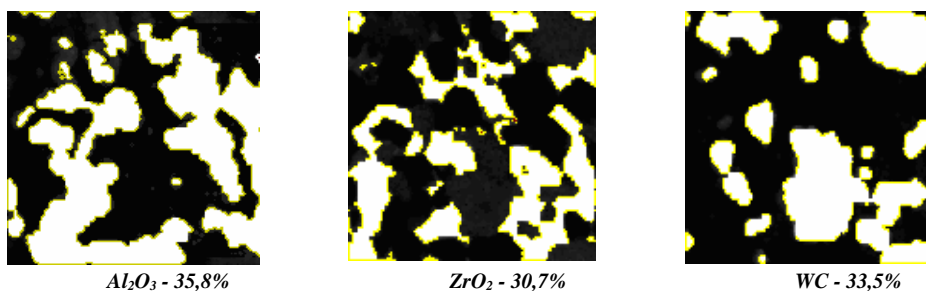
**Fig. 1** presents a set of high angel angular dark field (HAADF) image and corresponding elemental mapping images recorded in the STEM mode on a marked area.



**Fig.1** HAADF image and corresponding elemental mapping images recorded along marked area.

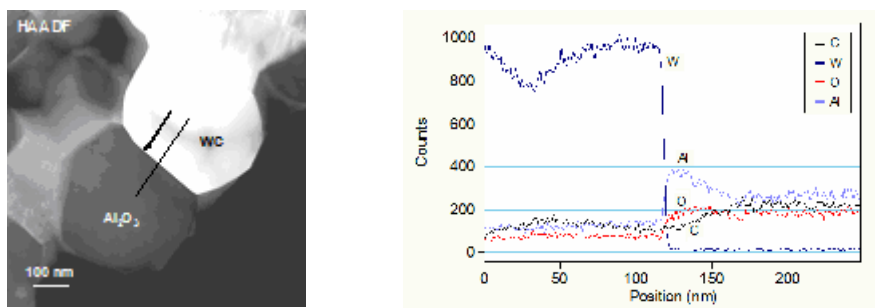
The obtained contrast is due to a different atomic number of elements ( $Z^2$ -contrast) and it allows to analyze a particulate phases. Taking to the account a distribution of Al, Zr and W, we can

calculate the mean grain size of  $\text{Al}_2\text{O}_3$ ,  $\text{ZrO}_2$ , WC phases, respectively. In this purpose the image analyze software Gatan Digital Micrograph™ 3.10.0 was used. The procedure of analyze was as follow: the elemental mapping images of Al, Zr and W were filtered applying "maximum" filter, then from obtained binary images the histograms of gray level were created and determination the threshold for maximum intensity of particulate phases were performed. Finally, we obtained the images of particulate phases and measured values of their mean grain size and filled area (**Fig. 2**). The mean grain sizes of investigated phases were as follow:  $\text{Al}_2\text{O}_3$  – 330 nm,  $\text{ZrO}_2$  – 280 nm and WC – 340 nm, and relative area fraction: 35.8, 30.7, and 33.5%, respectively. The obtained data were in a good agreement with these used as initial conditions for preparing of compacts (size of powders and volume proportion of each phase). Calculated mean grain size values for composite components are comparable with these calculated using much smaller image magnifications presented in [13].



**Fig.2** Results of images analyze showing distribution of particulate phases in the area indicated in Figure 1.

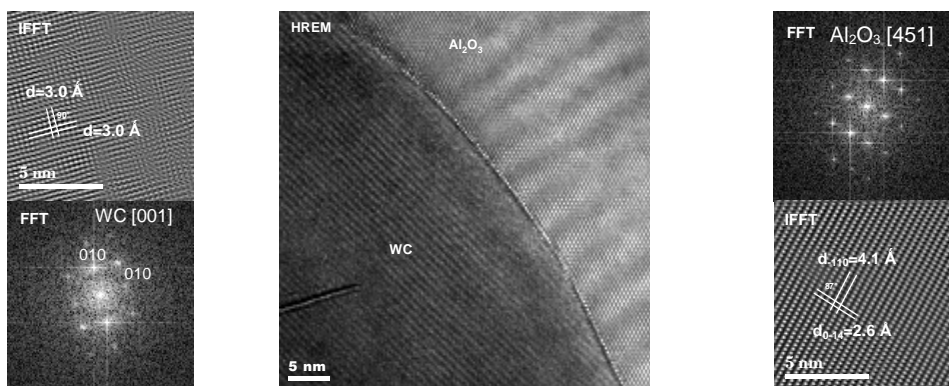
In order to analyze the grain boundaries between particulate phases a line scan analyses were performed for several grains. **Fig. 3** presents result of line scan analyse between WC and  $\text{Al}_2\text{O}_3$  grains performed along the marked line.



**Fig.3** HAADF image of composite microstructure and result of line scan analyze across the WC and  $\text{Al}_2\text{O}_3$  grains boundary.

One can see that both Al and W concentration is changing very sharp close the boundary, it means that no diffusion of this elements occurs. Concerning concentration changes of O, and C, is difficult to say about their diffusion, due to contamination effect, limit of detection of light elements by EDS method, thickness of sample and finally necessity of covering by the carbon nonconductive ceramic sample. Additionally, there is no information about the thickness of grains, therefore the overlapping of signal from different grains can occurs.

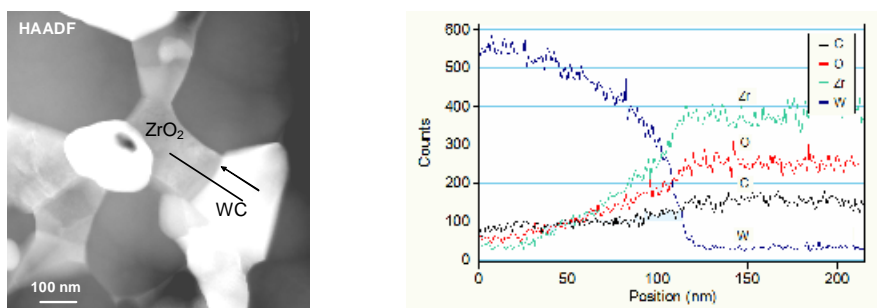
The observations of grain boundaries were performed in atomic scale also. **Fig. 4** presents the high-resolution transmission electron microscopy (HRTEM) showing grains boundary between WC and  $\text{Al}_2\text{O}_3$ .



**Fig.4** HRTEM image and corresponding FFT and IFFT images showing the grains boundary between WC and  $\text{Al}_2\text{O}_3$ .

One can see the sharp boundary between [001] and [451] oriented grains of WC and  $\text{Al}_2\text{O}_3$  phases, respectively. The inverse fast Fourier transform (IFFT) images showed that  $\text{Al}_2\text{O}_3$  is almost free from structural defects, whereas plenty of dislocations and stacking faults on {010} planes can be distinguish in WC grain.

**Fig. 5** presents a result of line scan analyse between WC and  $\text{ZrO}_2$  grains performed along the marked line.

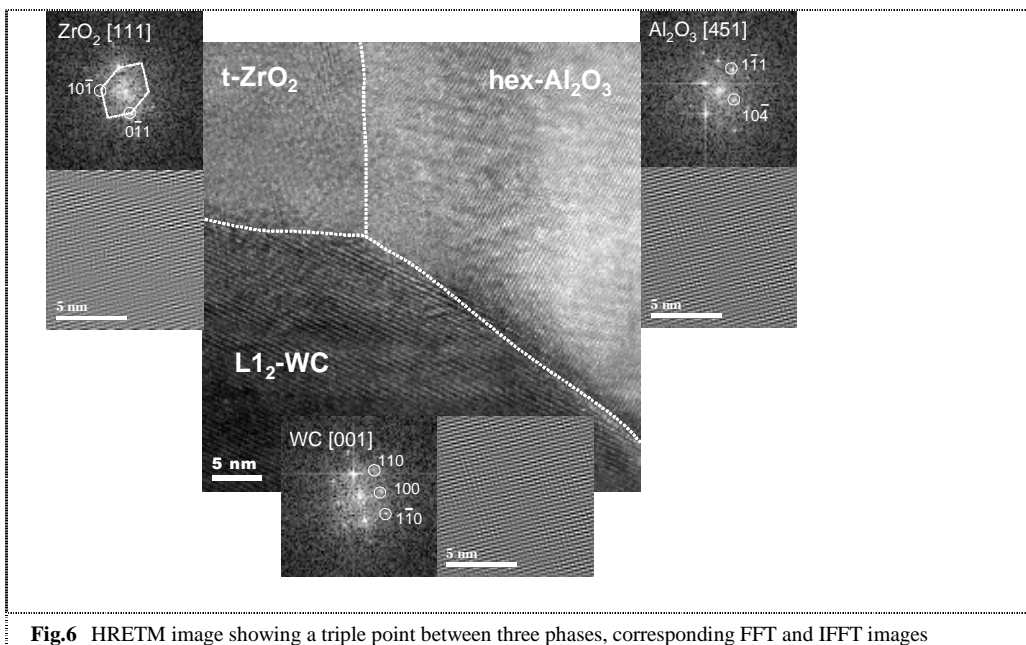


**Fig.5** HAADF image of composite microstructure and result of line scan analyze across the WC and  $\text{ZrO}_2$  grains boundary.

One can see that signal originating from Zr increases in measurement range from 25 nm up to about 110 nm reaching maximum inside the  $\text{ZrO}_2$  grain. This behaviour can be explain as both overlapping of the WC and  $\text{ZrO}_2$  grains or some diffusion of Zr into WC grain. However taking to the account character of signal originating from O (almost the same as in case of Zr) most probably this first one explanation is correct. The above assumption that the grain boundaries in  $\text{Al}_2\text{O}_3$ - $\text{ZrO}_2$ -WC composite are free from additional phases was proofed by high resolution observation.

Despite of chemical reaction, during high temperature treatment, some chemical elements segregation could occurs. In the WC grain the W/C and W/O ratio was changing continuously. It suggests that during sintering oxygen diffuse into the carbide structure. Parallel, carbon changes its content in carbide grain what has to change locally WC stoichiometry. It was previously observed in [14].

**Fig. 6** presents HRETM image showing a triple point between three phases. Each type of interphase boundaries is represented  $\text{Al}_2\text{O}_3\text{-ZrO}_2$ ,  $\text{Al}_2\text{O}_3\text{-WC}$  and  $\text{ZrO}_2\text{-WC}$ . The corresponding fast Fourier transforms (FFT) allowed to identify the crystallographic structure of coexistence phases. The tetragonal  $\text{ZrO}_2$  with [111] orientation,  $\text{L1}_2$  WC with [001] orientation and hexagonal  $\text{Al}_2\text{O}_3$  with [451] orientation were identified respectively by measurement of distances and angles on corresponded FFT's images. The phase boundaries areas did not show additional phases inside.



**Fig.6** HRETM image showing a triple point between three phases, corresponding FFT and IFFT images

#### 4 Conclusions

Investigations showed that the composite microstructure consisted of sub-micrometric grains of each phase randomly and homogeneously distributed. The grains boundaries were distinct and clear. This confirms high purity of used starting powders and a lack of chemical reactions between constituent phases at applied sintering conditions. HREM observations of interphase boundaries and the corresponding fast Fourier transforms (FFT) and inverse fast Fourier transforms (IFFT) allowed to identify the crystallographic structure of coexistence phases in boundaries area.

Local changing of the chemical composition of WC phase in contact with zirconia grains was detected. However, its nature should be elaborate in details, distinct changes of O/W and C/W ratios suggest that in the area of WC/zirconia interphase boundary chemical changes occurred.

WC in contact with alumina was stable. There were no distinct chemical changes on the WC/alumina interphase boundary.

### Acknowledgements

*The work was supported by the Polish Ministry of Science and Higher Education under project number 11.11.160.364.*

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